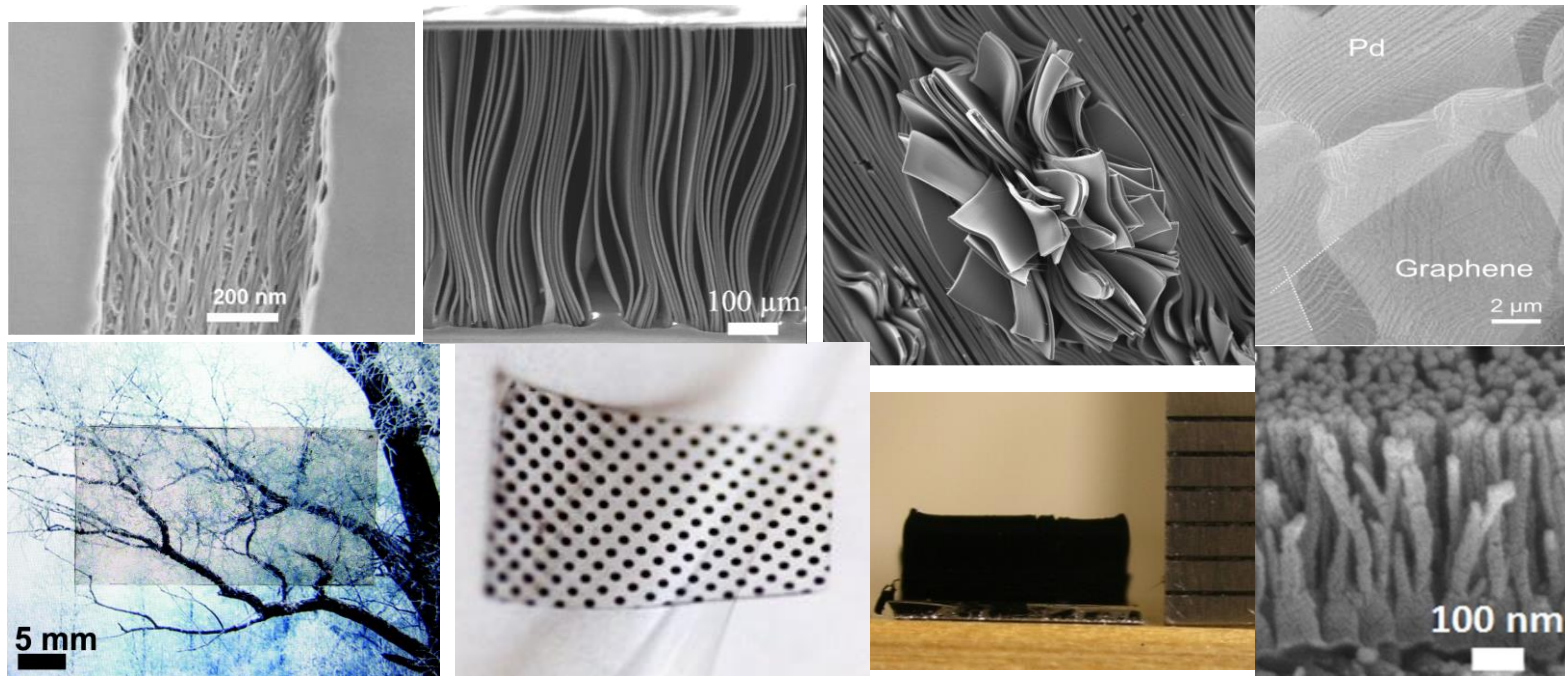


# Engineering Carbon Nanostructures and Architectures for High Performance and Multifunctional Electrodes

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# High Performance and Multifunctional Electrodes

## Multifunctional Electronics



Flexible and transparent smart phone, flexible computer, Flexible electronic newspaper (Wearable Electronics)

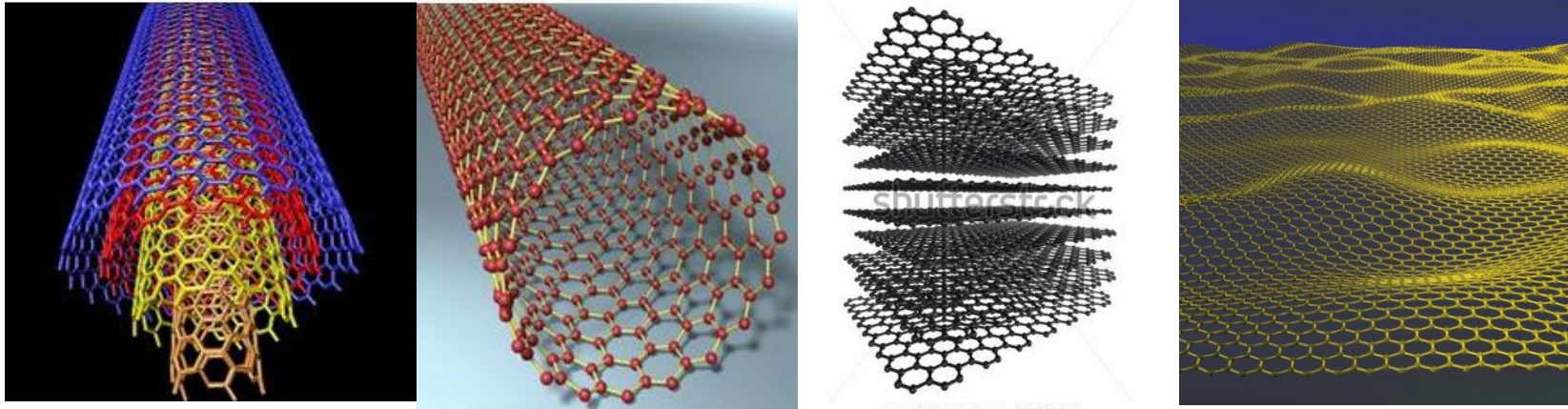


Devices require mechanically flexible, functional, and high performance energy storage systems

### Electrode Materials

- 3D Nanostructures
- Electro-Mechanical Stability
- Optical Transparency etc.

# SP<sup>2</sup> Carbon Nanostructured Materials



## ➤ **Mechanical Properties**

- *Strong sp<sup>2</sup> Carbon-Carbon covalent bonding*
- *High elastic modulus (1 TPa) and High strength*

## ➤ **Electrical and Optical Properties**

- *High Mobility*
- *Highly conductive w/wo mechanical deformation*
- *High current density (10<sup>9</sup> A/cm<sup>2</sup>)*
- *Optically Transparent*

## ➤ **In-plane Properties of Graphitic Carbon**

- *Good thermal conductivity (<3000W/mK)*
- *Good chemical stability*

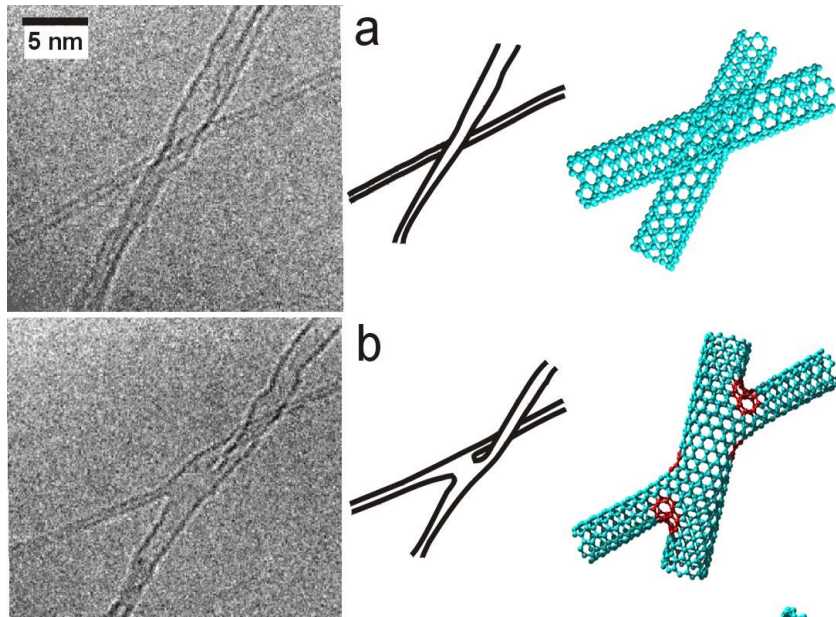


# Engineering SP2 Nanostructure

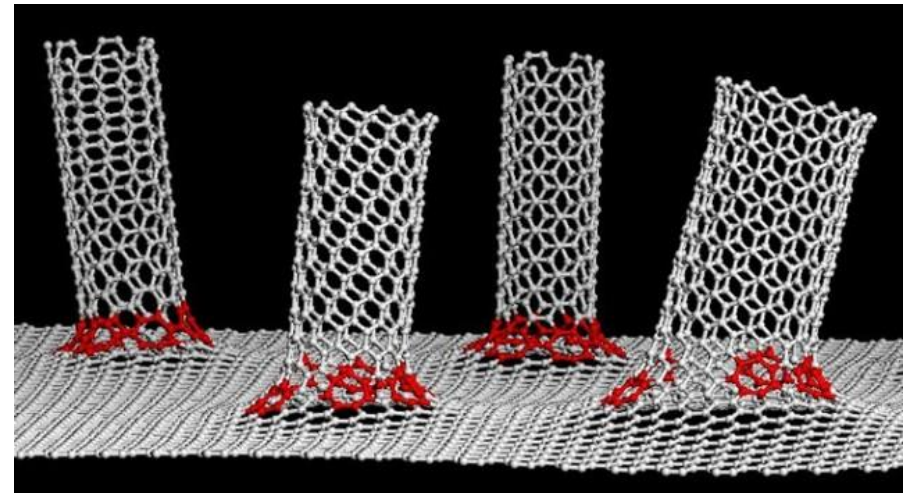
## Limitation of current CNT/graphene based networks

- ❑ Built on weak van der Waals interactions between CNTs, CNTs-Graphene
- ❑ Lower mechanical strength, electrical and thermal conductivities due to a lower pulling resistance, electron and phonon scatterings at these “unconnected” junctions

## Transforming physical Junctions into covalently bonded $sp^2$ Chemical Junctions



Terrones, Ajayan et al., PRL, 2002

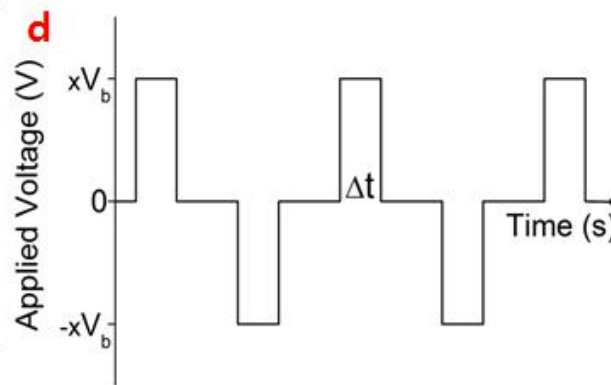
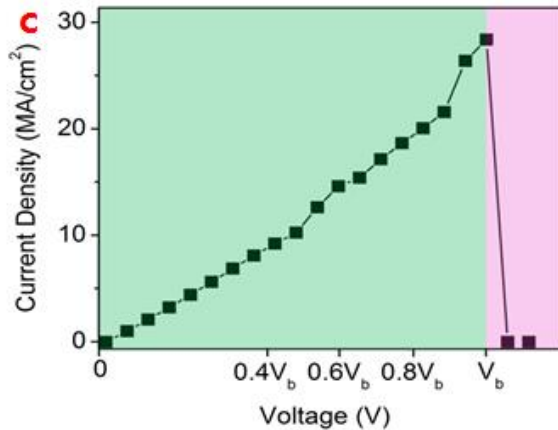
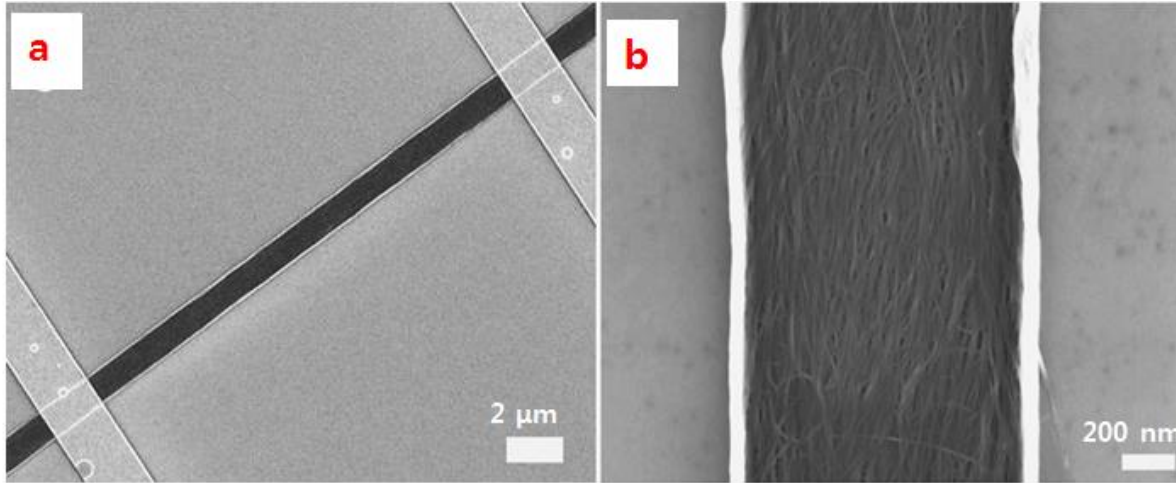


J. Tour et al., Nature Communications, 2012

# Engineering SP2 Nanostructure

## Restructuring $sp^2$ Lattice and Network Structure

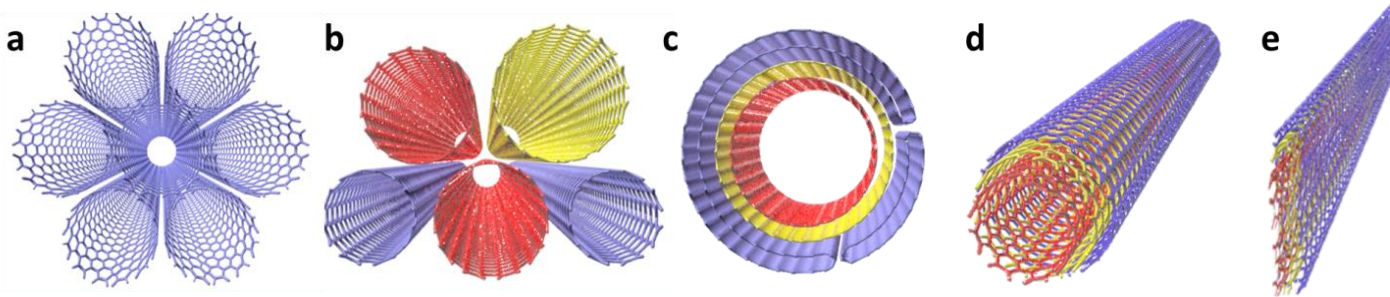
A voltage-induced *electrical fusion* of SWCNTs



H. Jung et al, Nature Communications, 2014

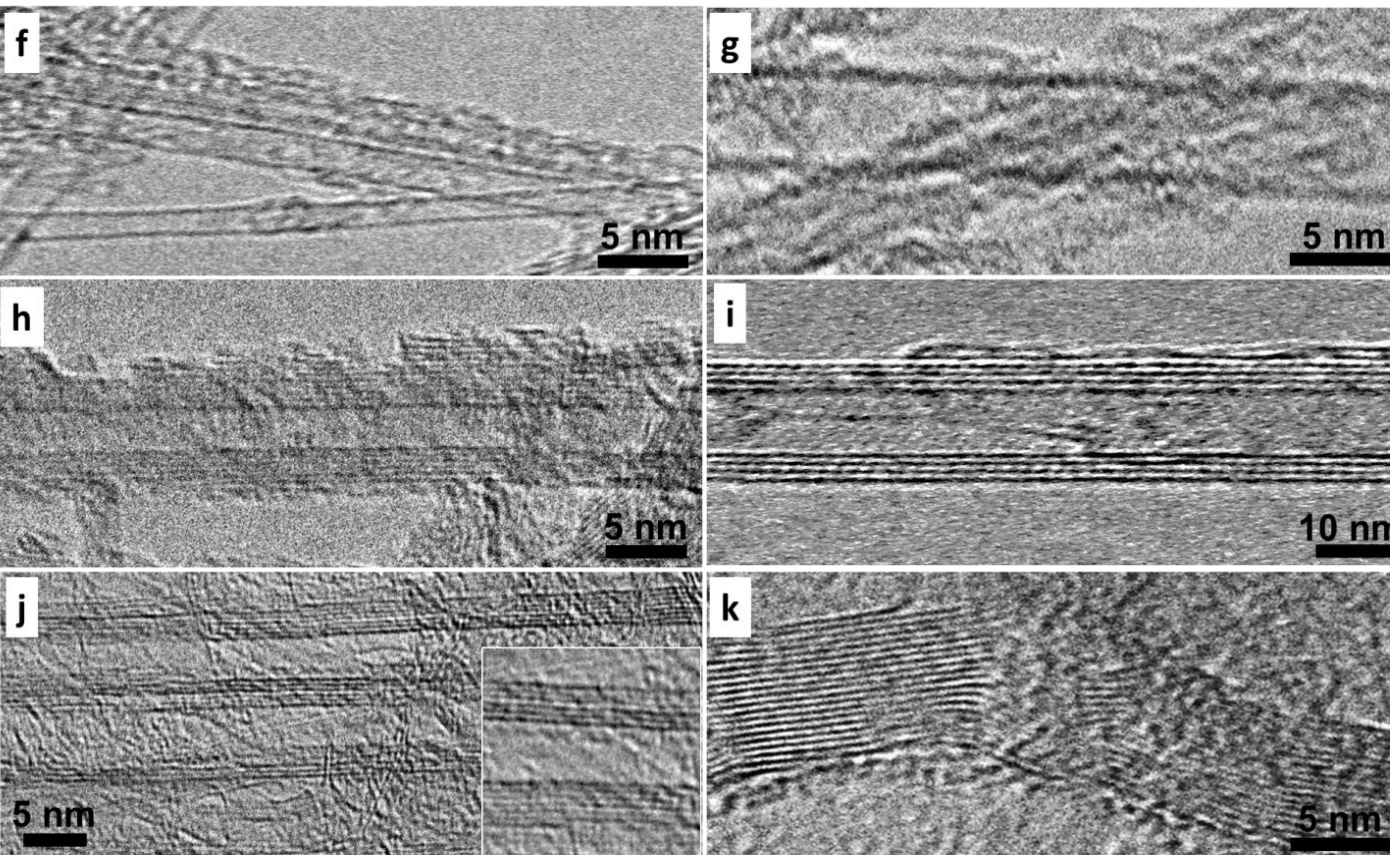


# Engineering SP2 Nanostructure

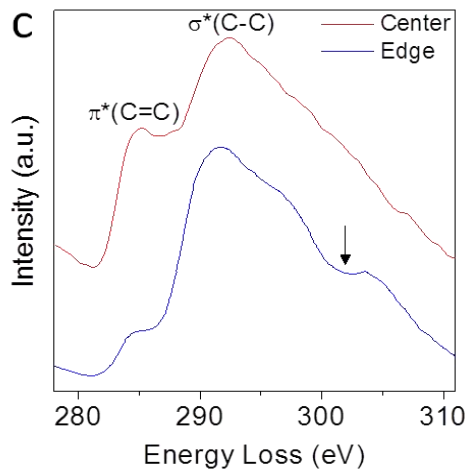
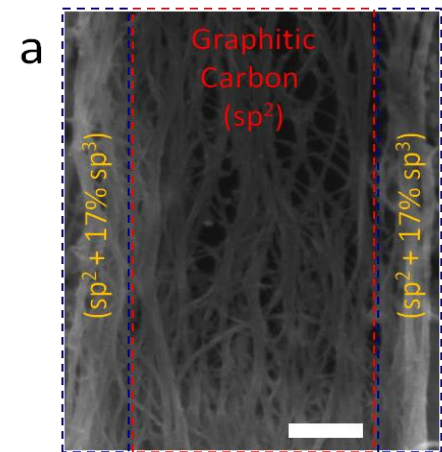
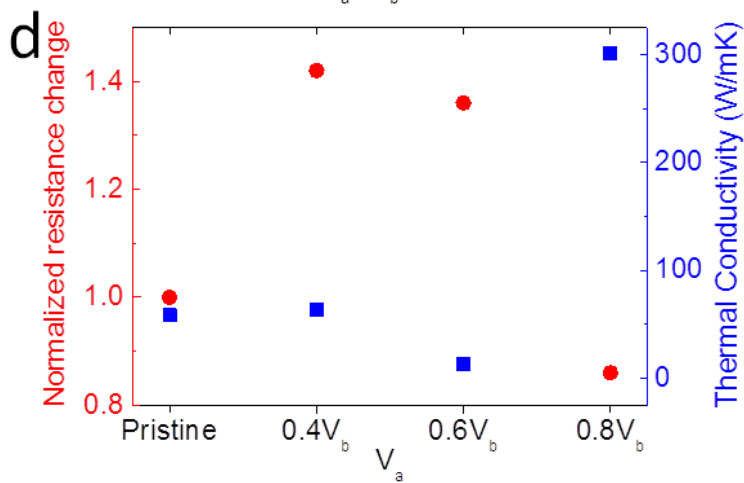
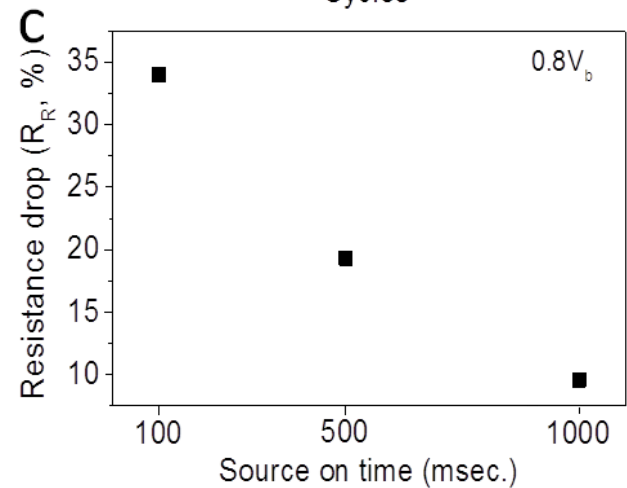
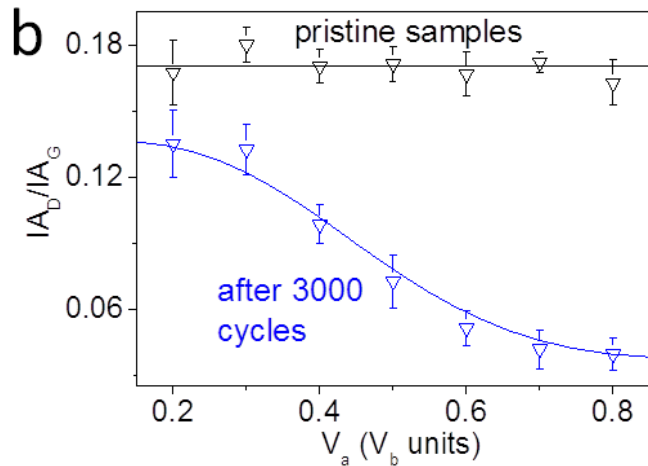
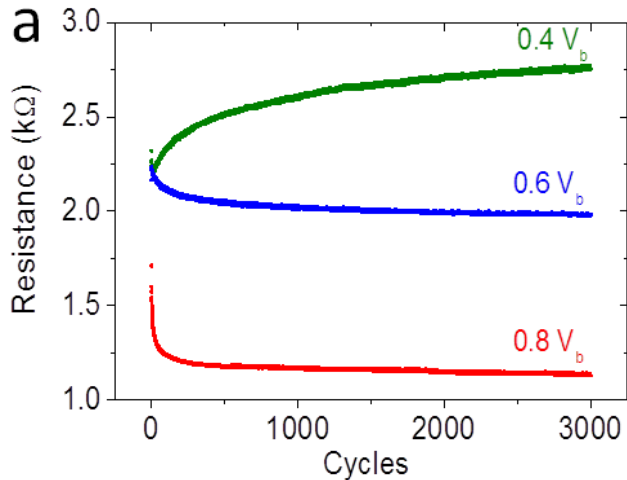


*Restructuring  
 $sp^2$  Lattice  
Structure*

*H. Jung et al,  
Nature Communications  
2014*

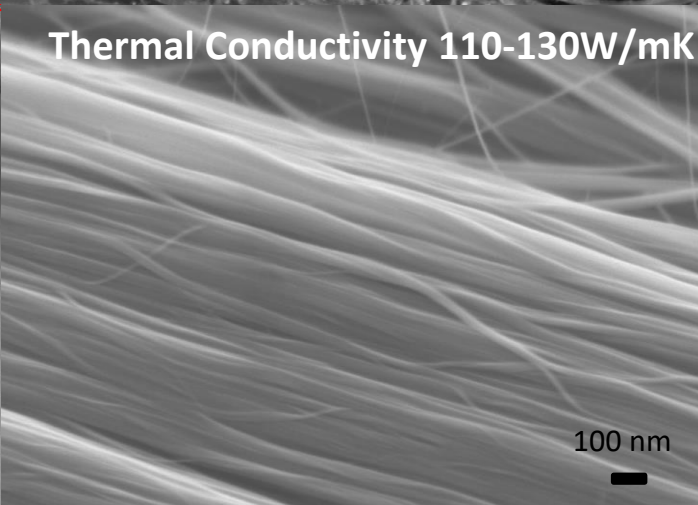
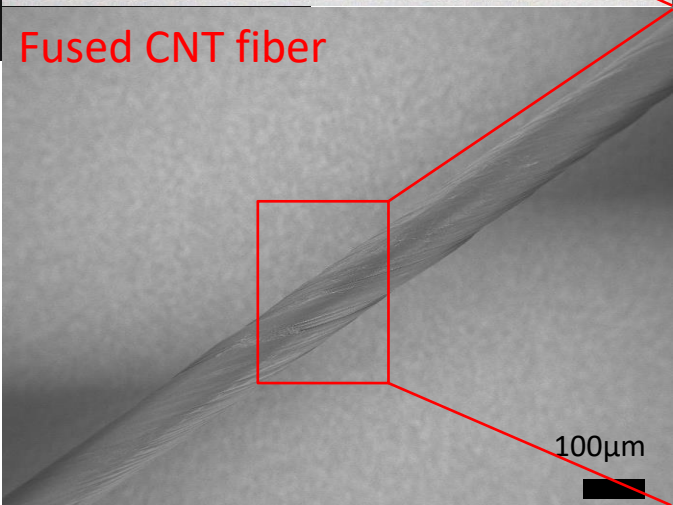
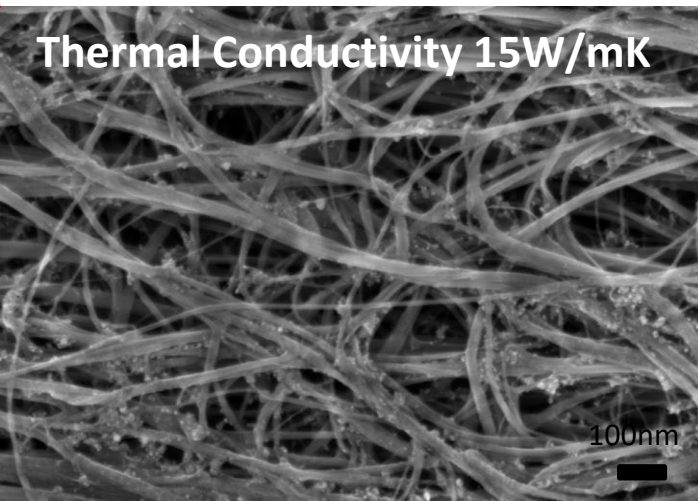
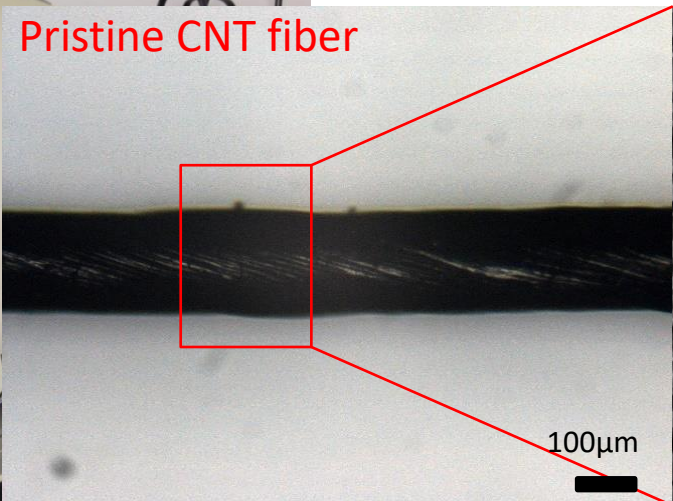


# Engineering SP2 Nanostructure





# Engineering Nanostructure and Morphology



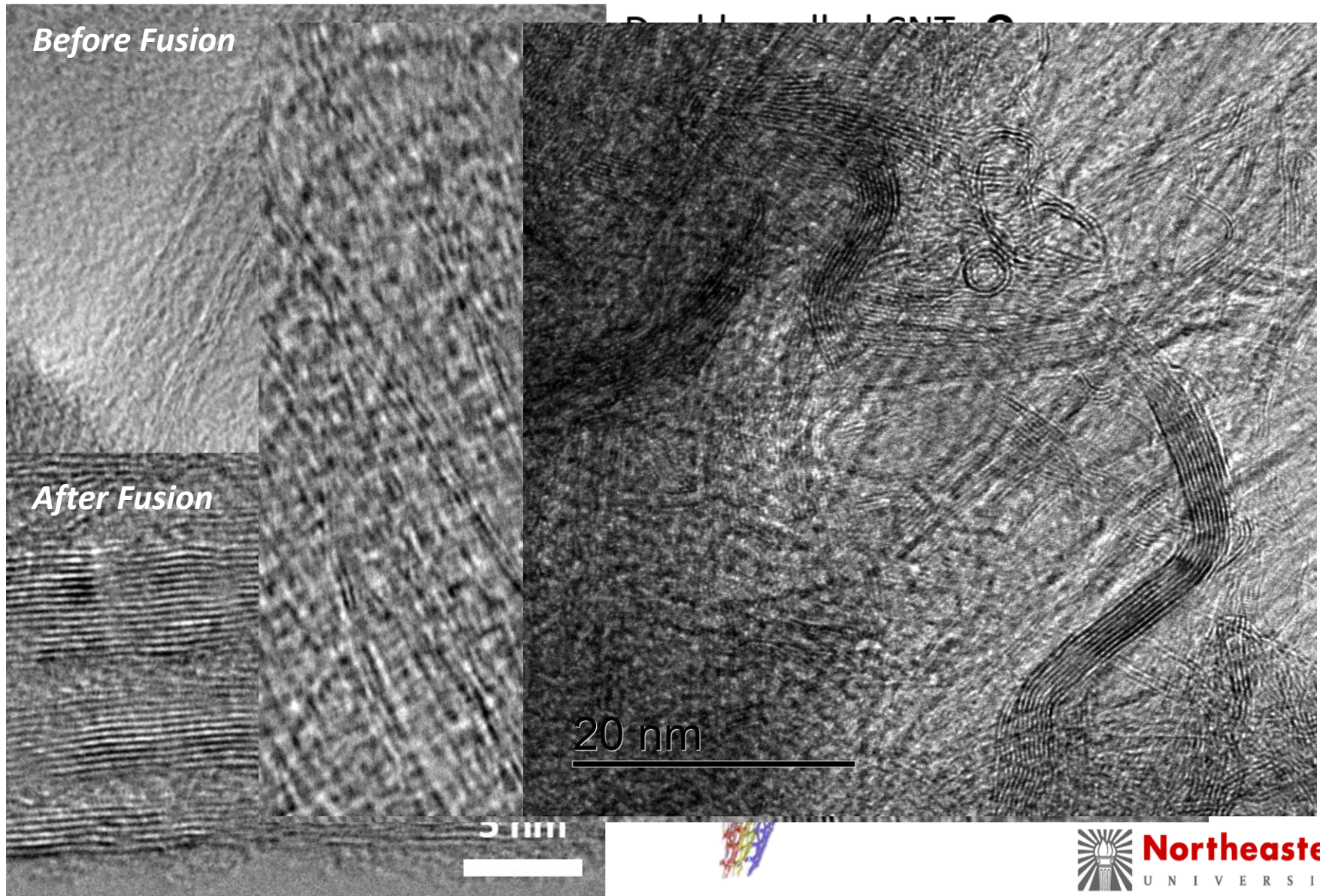
Supported by  
NSF-DMREF Program  
(Materials Genome Initiative)



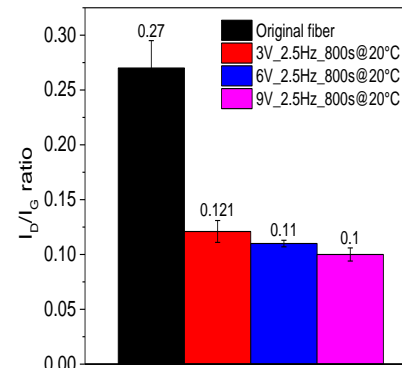
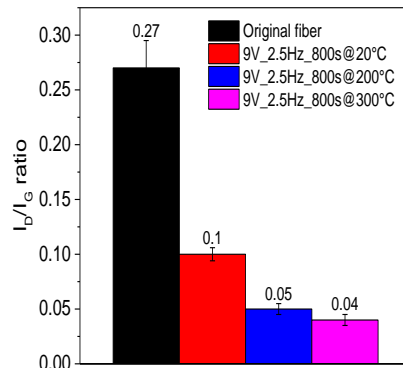
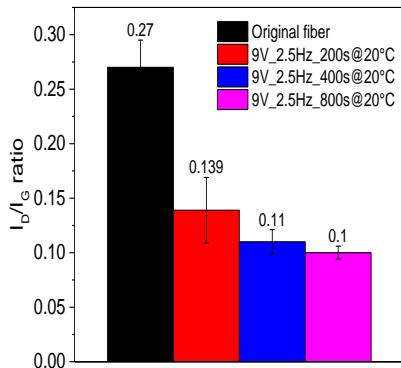
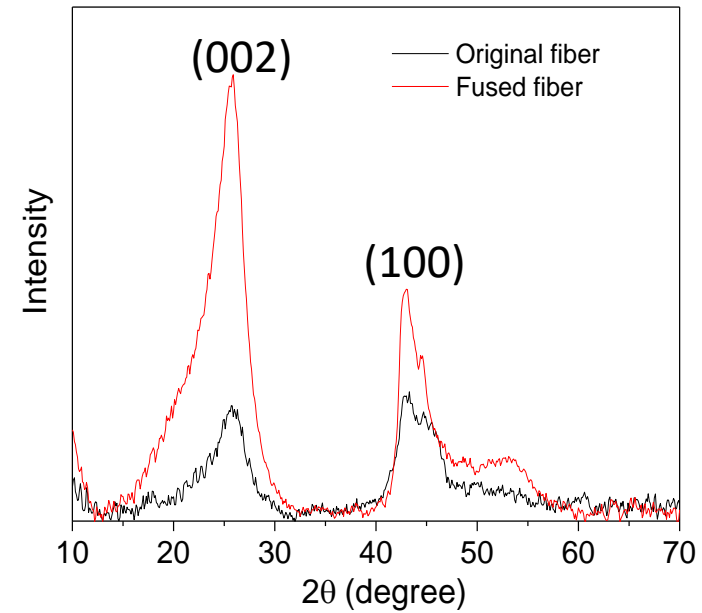
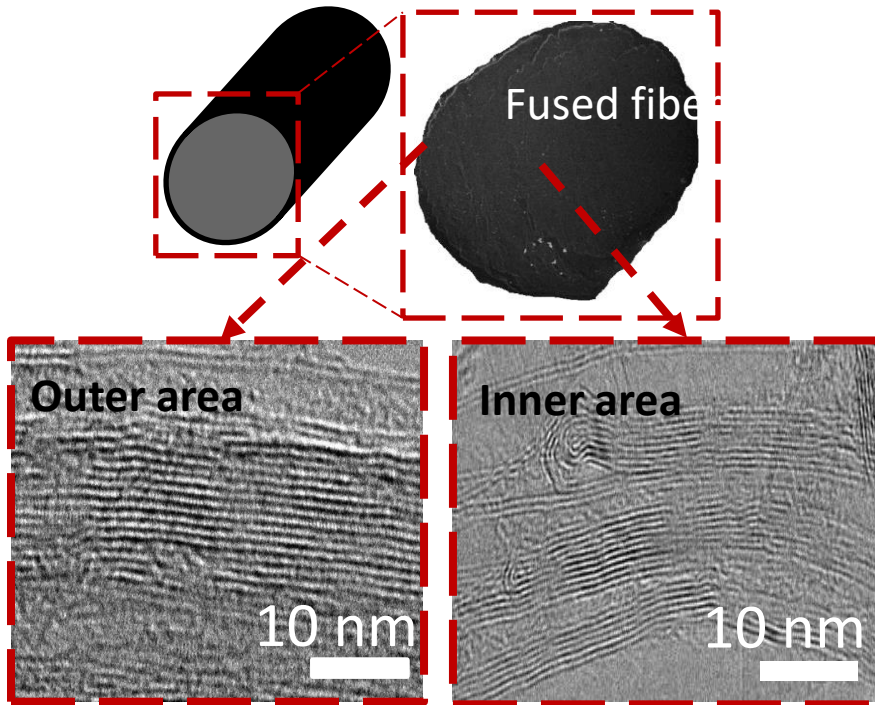
*J. Hao et al, Unpublished*



# Engineering Nanostructure and Morphology



# Engineering Nanostructure and Morphology





# Engineering 3D Nanoscale Architecture

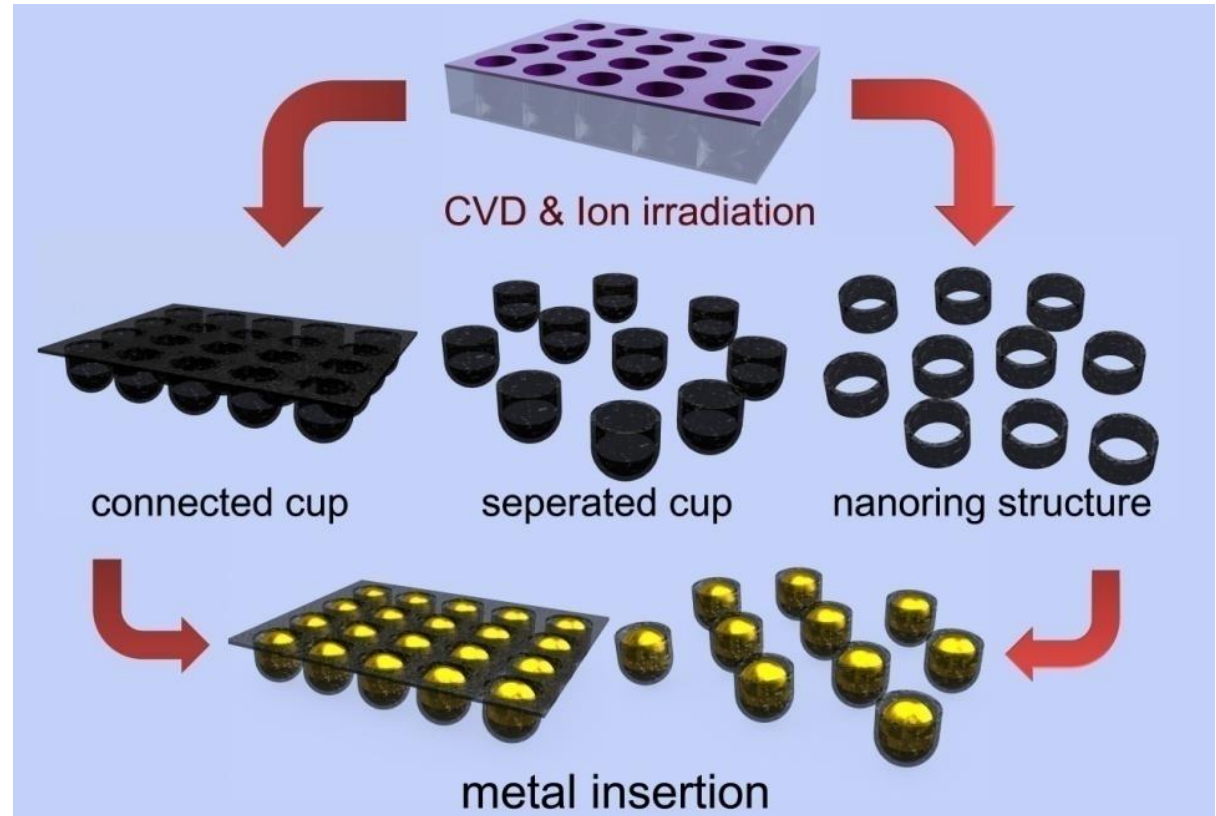
## Carbon Nanocups

### Graphitic nanostructures

having smaller length/diameter (L/D) aspect ratio, **nanoscale cup morphology**, can effectively contain other nanomaterials and polymers, leading to multi-component hybrid nanostructures.

### Multifunctional Nanosystems

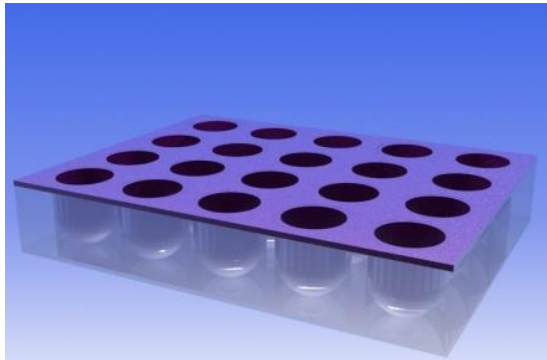
- Energy Storage
- Nanogram Quantity Container
- Multifunctional Sensors



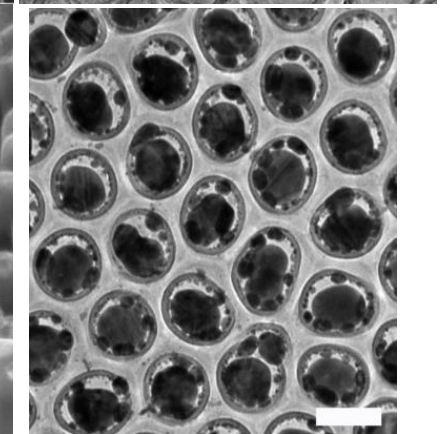
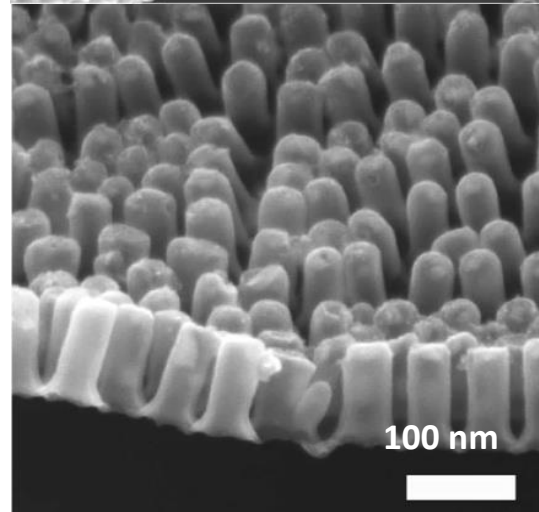
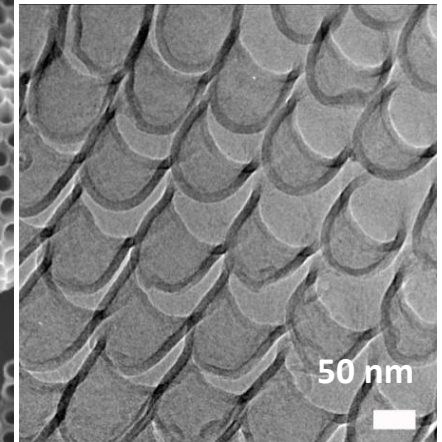
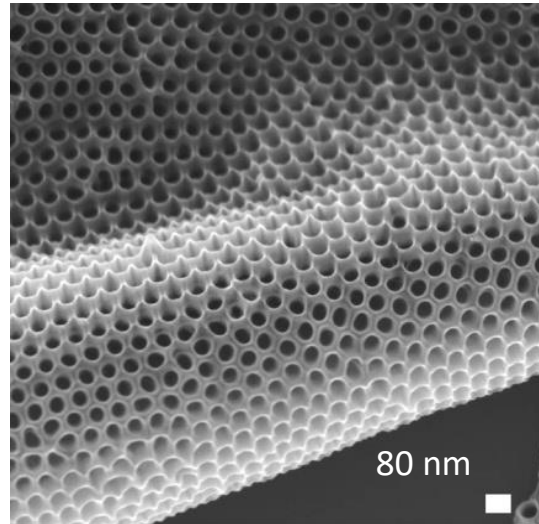
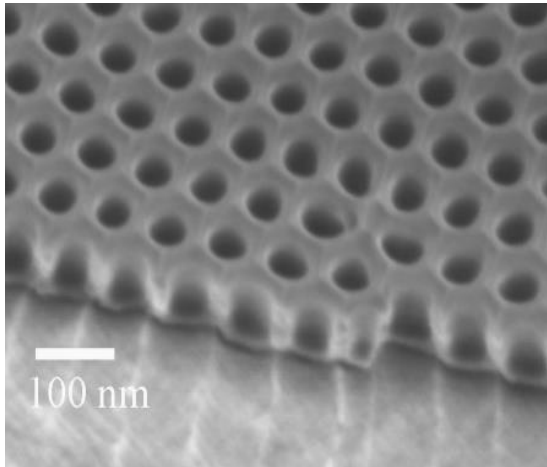
# Engineering 3D Nanoscale Architecture

## **Fabrication Process**

*The length of nanochannels are controlled by second anodizing time.*



**Thermal  
CVD  
of Carbon**



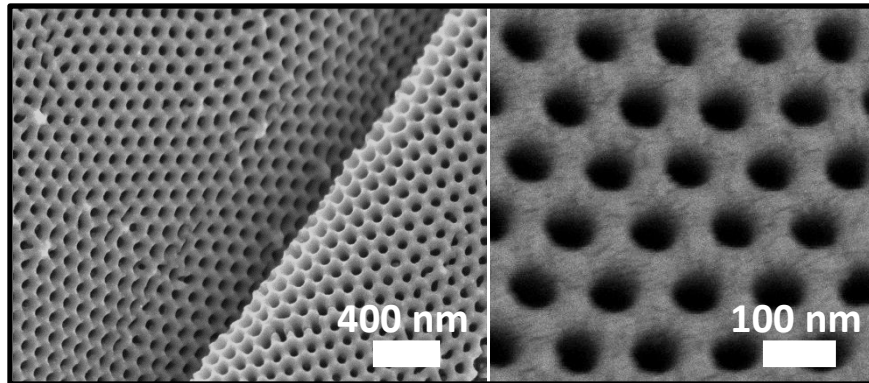
*H. Chun, et. al., ACS Nano (2009)*

*H. Jung et al. Scientific Reports (2011)*

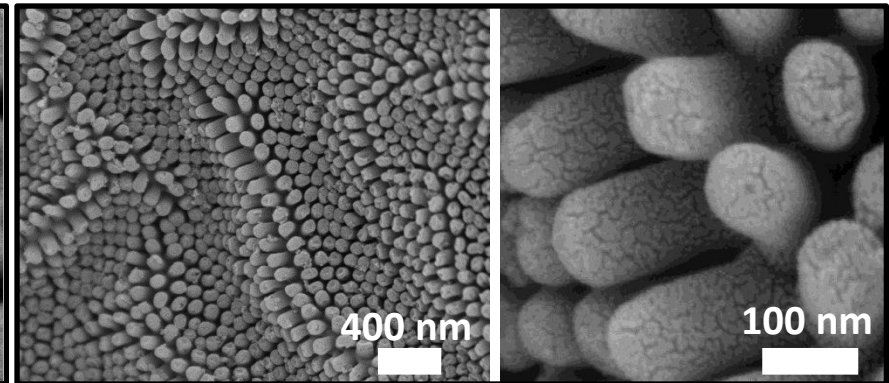


# Engineering 3D Nanoscale Architecture

TOP view



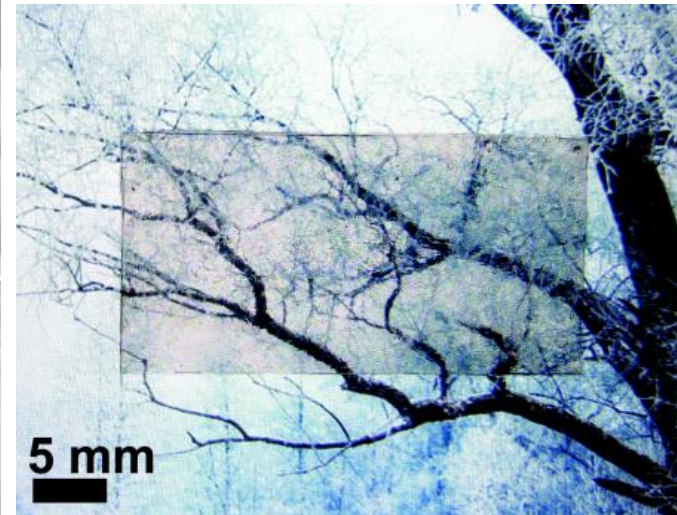
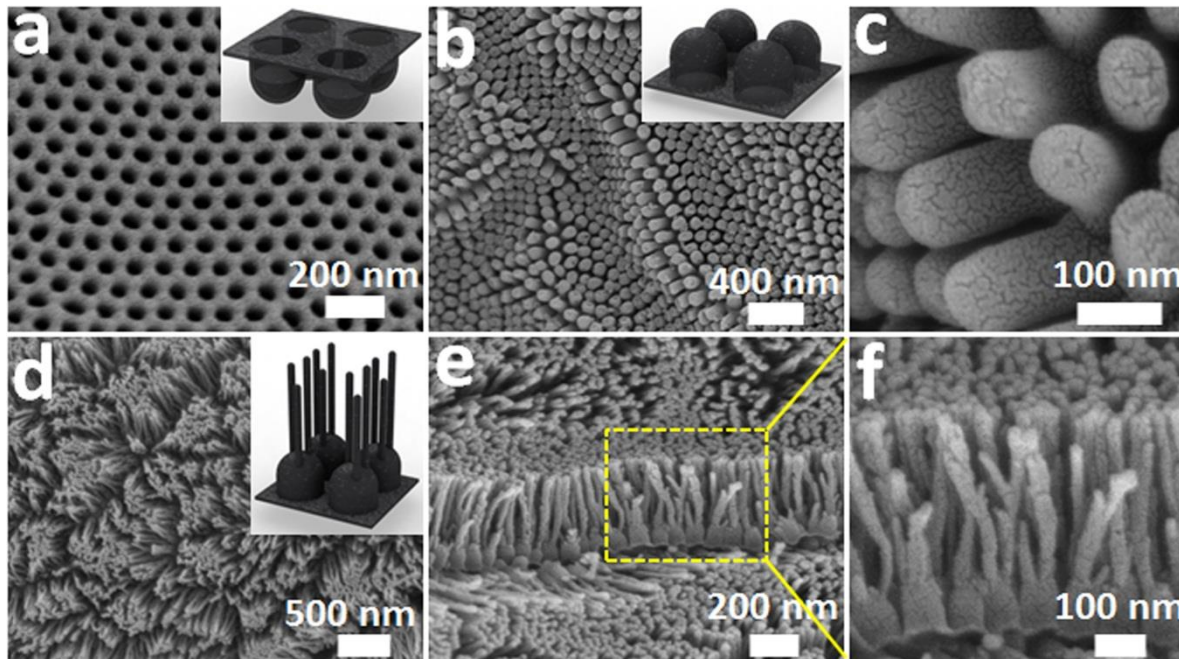
BOTTOM view



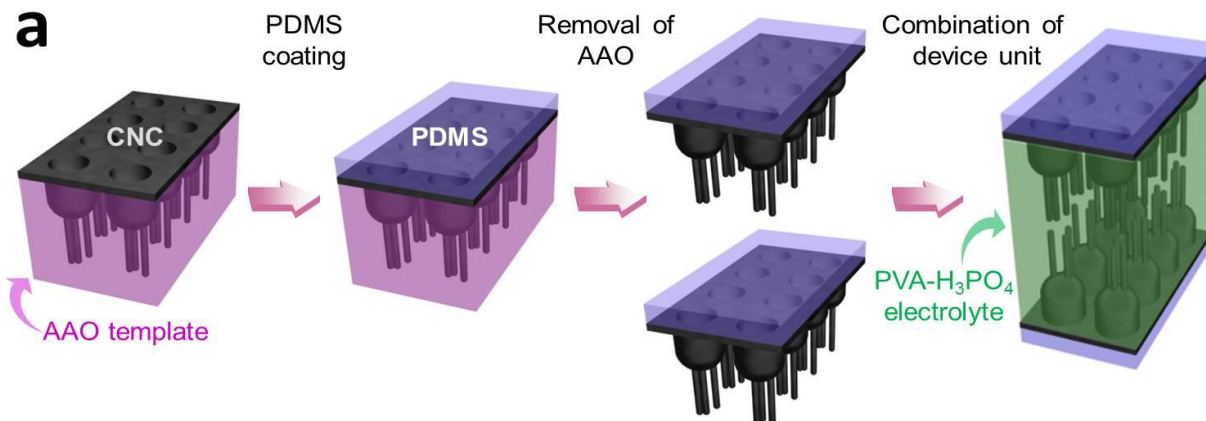
## 3D Carbon Nanostructured Film for Supercapacitor Electrodes

- Electrically Conductive: Surface Conductivity: 117 S/m
- High surface area and highly disordered graphitic layers provides the effective permeation of the polymer electrolyte and their conformal packaging with electrodes.
- Unique nanoscale cup feature enables the easy access and faster transport of ions at the electrode/electrolyte interface resulting in higher power capability.
- High current carrying capability, substantial mechanical strength, and small effective electrode thickness (5-10 nm: *80-85% Transmittance at 550nm wavelength*) allow us to build *optically transparent* and *mechanically flexible* reliable thin-film (solid state) energy storage devices.

# Flexible and Transparent Supercapacitors



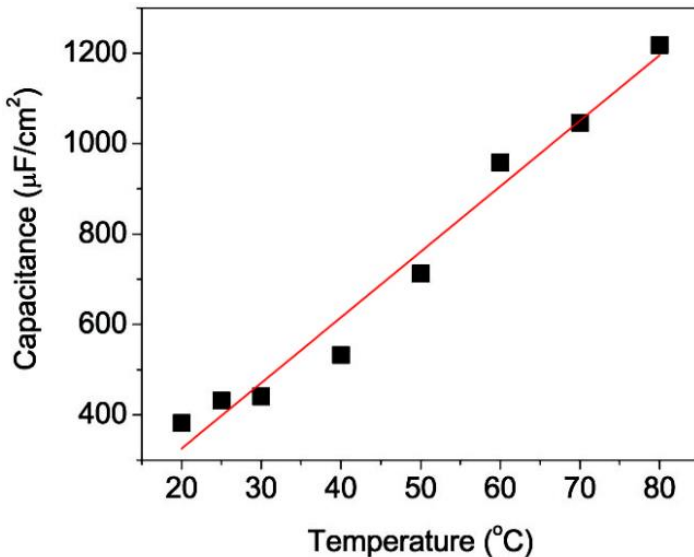
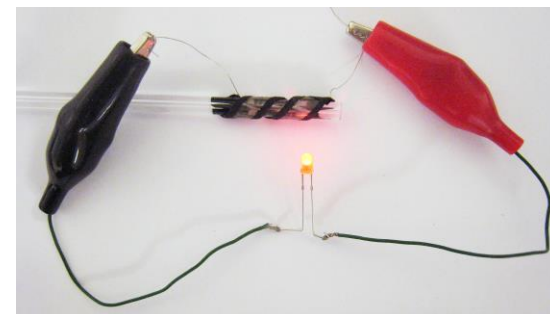
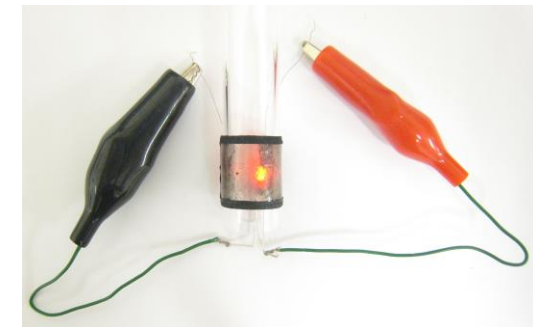
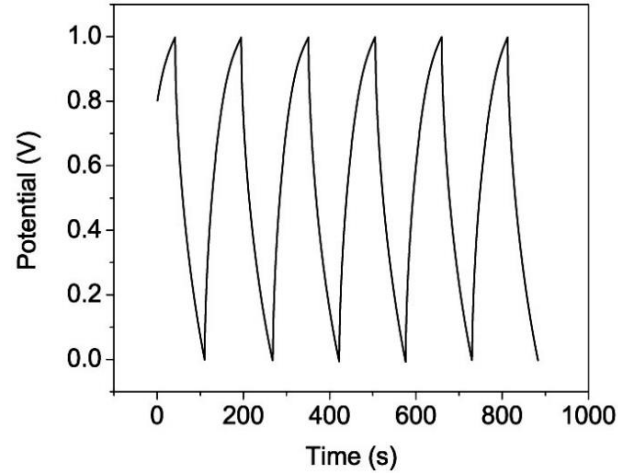
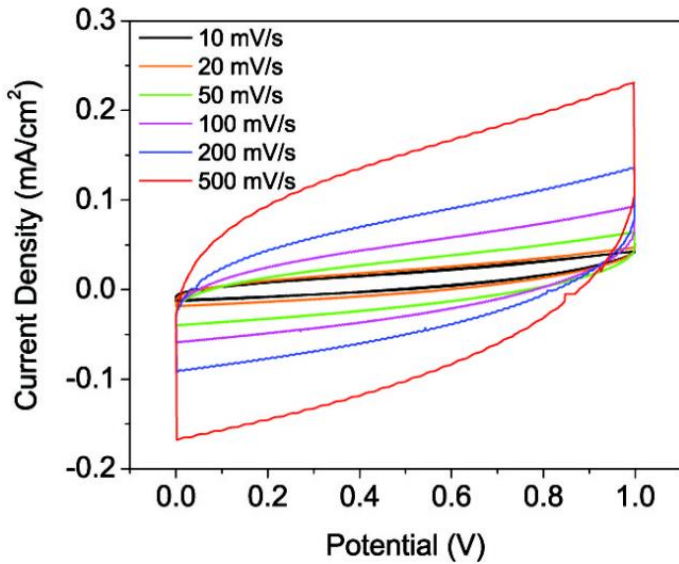
(a) concave and (b, c) convex and (d-f) branched nanocup films (*H. Jung et al. Scientific Reports 2012*)



- ❑ CNC films: Outer graphene layers are acting as current collectors and the innermost layer exposed electrolyte is acting as an electrode.
- ❑ Polymer electrolyte (PVA-H<sub>3</sub>PO<sub>4</sub>) is acting as both electrolyte and separator.

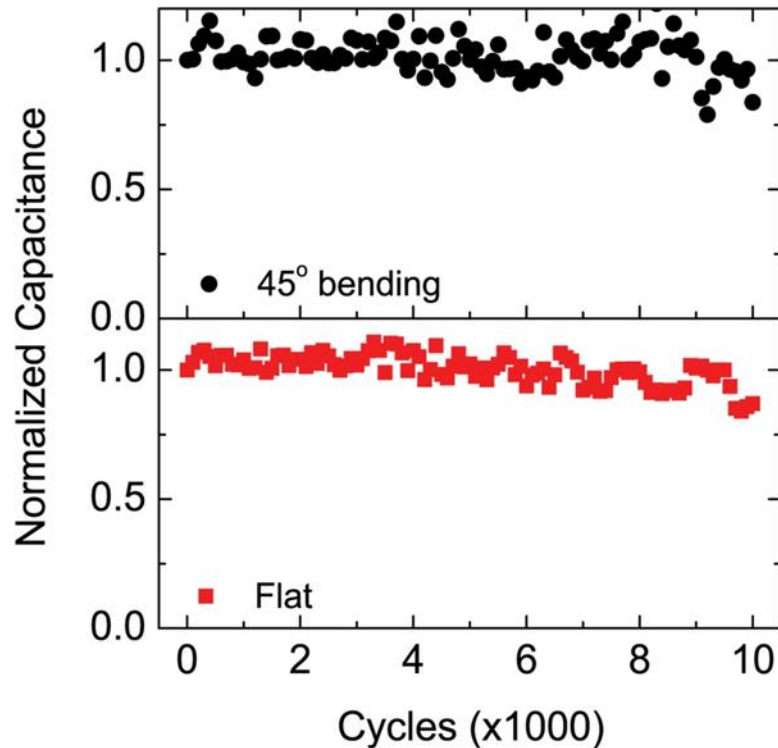


# Flexible and Transparent Supercapacitors

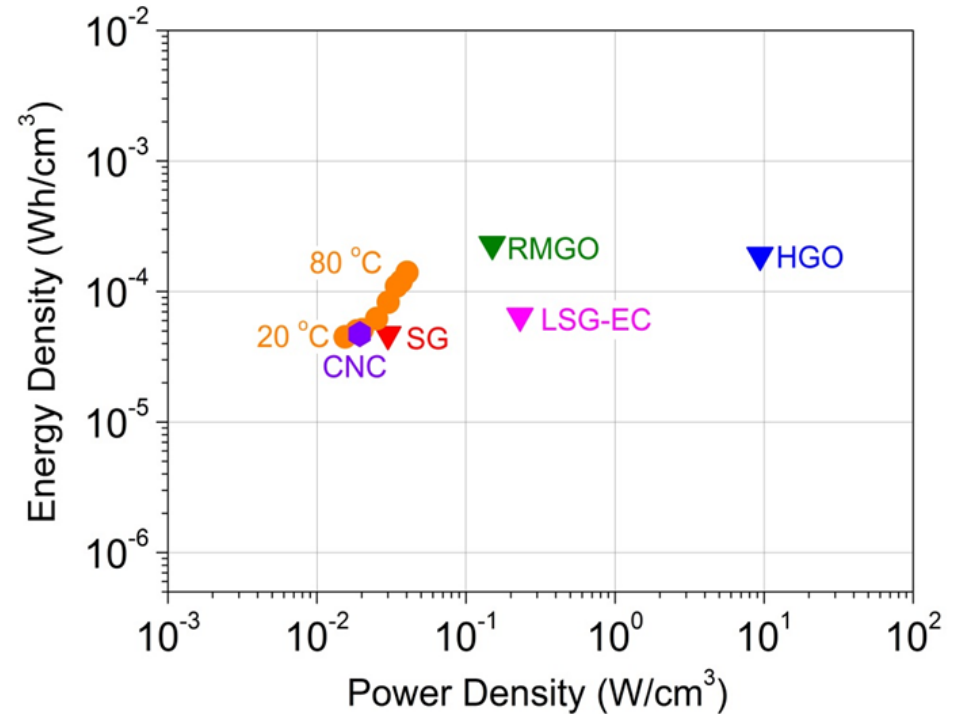


(a) Cyclic voltammetry (CV) measured with 10 – 500  $\text{mVs}^{-1}$  scan rates. (b) Galvanostatic charge/discharge (CD) results measured at a constant current density of  $5 \mu\text{A}/\text{cm}^2$ . The capacitances by the geometrical area calculated from CD curves are  $409 \mu\text{F}/\text{cm}^2$ . (c) The capacitance change as a function of temperature

# Flexible and Transparent Supercapacitors



Normalized capacitance as a function of cycle-number (10,000) and w/o the mechanical deformation (45° bending).



(SG: single layer graphene, RMGO: reduced multilayer graphene oxide, HGO: hydrated graphitic oxide, LSG-EC: laser-scribed graphene electrochemical capacitor)



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